Erratum

Polymerization of Olefins through Heterogeneous Catalysis—V. Gas-Liquid Mass Transfer Limitations in Liquid Slurry Reactions

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[article in J. Appl. Polym. Sci., 32, 5451-5479 (1986)]

Equation (8) on page 5454 read:

$$k_l a = 9.85 \times 10^{-4} D_l^{0.6} N^{1.2} g^{-0.053} v_l^{-0.45} Q^{0.76} D_T^{-0.72} (F_1 F_2)^{-0.6}$$
 (8)

It should read:

$$k_l a = 9.85 \times 10^{-4} D_I^{0.6} N^{1.2} g^{-0.053} v_l^{-0.45} Q^{0.36} D_T^{-0.72} (F_1 F_2)^{-0.6}$$
 (8)

Equation (10) on page 5454 read:

$$a_o = 230 \frac{(P/V)^{0.4} \rho_l^{0.2}}{\sigma_l^{0.6}} v_s^{1/2} \tag{10}$$

It should read:

$$a_o = 0.280 \frac{(P/V)^{0.4} \rho_l^{0.2}}{\sigma_l^{0.6}} v_s^{1/2}$$
(10)

Equation (18) on page 5457 read:

$$\frac{P_g}{P_{ug}} = 0.2296 \left(\frac{P_{ug}^2 N D_I^3}{Q^{0.56}} \right)^{0.45} \tag{18}$$

It should read:

$$P_g = 0.2296 \left(\frac{P_{ug}^2 N D_I^3}{Q^{0.56}} \right)^{0.45} \tag{18}$$

Equation 20 on page 5457 read:

$$N_{\rm sus} = \left[\left(\frac{D_T}{D_I} \right)^{4/3} d_p^{0.2} \mu_l^{0.1} g^{0.45} (\rho_p - \rho_l) / \rho_l^{0.05} D_I^{0.85} \right] (\phi_w \rho_s)^{0.13} \tag{20}$$

It should read:

$$N_{\text{sus}} = \left[\left(\frac{D_T}{D_I} \right)^{4/3} d_p^{0.2} \mu_l^{0.1} g^{0.45} \frac{(\rho_p - \rho_l)^{0.45}}{\rho_p^{0.68} D_I^{0.85}} \right] (\phi_w \rho_s)^{0.13}$$
 (20)



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